

**Saylor Foundation's  
ME 103 Assessment 1: Guide to Responding**

**Instructions:** Please answer each of the following questions to the best of your ability.

**Questions**

1. What are the important features of a property?
2. Define 1 standard atmosphere (atm) in SI unit.
3. Determine the force needed to lift up an 1 kg object.
4. A U-tube manometer filled with mercury is used to measure the pressure drop of air flow in a pipe. Estimate the pressure drop if the manometer shows a deflection of 5 mm. The density of mercury is  $13.6 \times 10^3 \text{ kg/m}^3$ .
5. What would be the change in the temperature of 1 kg of metal at the foot of a fall from an altitude of 5000 m above ground level, assuming that no energy is lost during the fall? The specific heat of the metal is 100 kJ/kg.
6. What are the common types of pressures that one often comes across?

**Solutions:**

1. At a particular state of the system, a property should be finite and well-defined, irrespective of how the system was brought to the state.

2.  $1 \text{ atm} = 1.01325 \times 10^5 \text{ N/m}^2$

3.  $F = m \times g = 1 \text{ kg} \times 9.8 \text{ m/s}^2 = 9.8 \text{ m/s}^2$ .

4. The pressure drop  $\Delta P$  is measured by the difference in the heights of the mercury columns (deflection  $\Delta h$ ) in the manometer as follows:

$\Delta P = \rho g \Delta h = 666 \text{ Pa}$ .

5. During the fall, potential energy of the metal is converted to thermal energy if no energy is lost during the fall. In other words, potential energy = heat to increase temperature of the metal.

Hence,  $m \cdot g \cdot h = m \cdot C_v \cdot \Delta T$ , or  $\Delta T = g \cdot h / C_v = 9.8 \times 5000 / 100 \times 10^3 = 0.49 \text{ K}$ . Change in temperature of metal is  $\sim 0.5$  degree Kelvin.

6. There are four types of pressures that one often comes across: atmospheric, absolute, gauge and vacuum.

