

## Saylor Foundation's

### ME103 Assessment, Unit 6: Guide to Responding

**Instructions:** Please answer each of the following questions to the best of your ability.

#### Questions:

1. Consider an air-standard Otto-cycle engine with a compression ratio of 8. At the beginning of compression, the pressure and temperature at the beginning of compression are 100 kPa and 300 K, respectively. Determine the thermal efficiency of the engine if 1.8 kJ is added per kg air and  $C_v = 0.7165$  J/kg

2. Determine the thermal efficiency of a Carnot steam cycle operating between a source temperature of 583 K and a sink temperature of 306 K. The boiler and condenser pressures are 10 MPa and 5 kPa, respectively.

#### Solutions:

1. Draw the Otto cycle.

- At State 1,  $P_1 = 100$  kPa and  $T_1 = 300$  K.
- At State 2,  $P_2 = P_1 \mu^k$  and  $T_2/T_1 = (P_2/P_1)^{k-1/k} = \mu^{k-1}$ , where  $\mu$  is the compression ratio and  $k$  is 1.4 for air. This gives  $P_2 = 1840$  kPa and  $T_2 = 689$  K.
- At State 3, heat is added to air at  $q_H = C_{v,m}(T_3 - T_2) = 1800$ . This gives  $T_3 = 3196$  K.  $P_3 = P_2(T_3/T_2) = 8524$  K
- At State 4, applying the Otto cycle equations, we obtain  $P_4 = 464$  kPa and  $T_4 = 1391$  K.

The thermal efficiency of the engine is:

$$\eta_{th} = 1 - q_L/q_H = 1 - C_v(T_4 - T_1) / q_H = 0.56$$



2. Draw the Otto engine cycle and define states 1-4 clearly.

Use steam tables to look up properties of different states along the vapor cycle.  
For states 1 and 4, water is saturated.

- At state 2,  $H_2 = 1407$  kJ/kg,  $s_2 = 3.36$  kJ/kg K
- At state 3,  $H_3 = 2725$  kJ/kg,  $s_3 = 5.61$  kJ/kg K
- At state 1,  $x_1 = 0.36$ ,  $H_1 = h_{f,1} + x_1 h_{fg,1} = 1020$  kJ/kg
- At state 4,  $x_1 = 0.65$ ,  $H_4 = h_{f,4} + x_4 h_{fg,4} = 1711$  kJ/kg

Thermal efficiency is given by  $N_{th} = w_{net}/q_H = (w_T - w_c) / q_H$

- turbine work  $w_T = h_3 - h_4$
- compressor work:  $w_c = h_2 - h_1$
- heat added:  $q_H = h_3 - h_2$

Hence, thermal efficiency  $N_{th} = w_{net}/q_H = (w_T - w_c) / q_H = 1 - (h_1 - h_4) / (h_3 - h_2) = 0.48$

